

P1 Crystals, glasses, and gels: controlling solid state forms of simvastatin

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Simvastatin (**Figure 1**) is one of the most widely used active pharmaceutical ingredients (API) for the treatment of hyperlipidemias. The compound is employed as a solid in drug formulations. As such, particular attention must be given to the characterization of its different crystal forms (polymorphs), their stability domains, and the nature of the phase transitions that relate them. Modifications in crystal packing and crystallinity are commonly accompanied by significant changes in physical properties (e.g. variations in solubility), and may lead to serious problems in terms of reproducible preparation and safe use of a medicine. As a result, the characterization of structure–energetics relationships for solid APIs is of considerable importance in the pharmaceutical industry.

In this work simvastatin will be used as a model to illustrate how experimental (single crystal X-ray diffraction, hot stage microscopy, differential scanning calorimetry, and solubility measurements) and molecular dynamics results can be combined to guide the control of API production in specific solid state forms. Simvastatin has three known polymorphs. Form I is the thermodynamically stable one under ambient conditions and converts to the other two by fast solid–solid phase transitions (at 275.2 K for the form I → form II transition and 236.9 K for form II → form III). [1,2] Furthermore, long-lived amorphous phases of simvastatin have also been observed. [3]

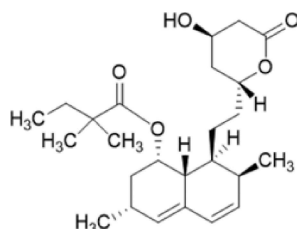


Figure 1. Molecular structure of simvastatin.

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